INFLUENCE OF CYCLODEXTRINS ON THE SODIUM BOROHYDRIDE REDUCTION OF CYCLOHEXENONES

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The sodium borohydride reduction of cyclohexenones in aqueous medium is changed in the presence of cyclodextrins. β -cyclodextrin favours the 1,4-reduction over the 1,2-reduction whereas α -cyclodextrin favours the 1,2-reduction.

Cyclodextrins form inclusion complexes with a wide variety of substrates. 1) Cyclodextrins and derivatives affect the rates of many kinds of chemical reactions and are used as enzyme models in biomimetic chemistry. 2) More recently, studies on the cyclodextrin reactions have been focused on their selectivity. 3) A recent report 4) prompts us to disclose our progress pertainting the sodium borohydride reduction of cyclohexenones in the presence of cyclodextrins. The following procedure is illustrative of the reaction: enone (2 mmol) was stirred in $100\ \text{mL}$ of distilled water in the presence of cyclodextrin (2 mmol) for 2 h before the reduction. The pH of distilled water was set up at 7 by addition of sodium hydroxyde prior to use as solvent. Sodium borohydride (4 mmol) in water was added dropwise and the reaction let to stand 24 h before analysis. The aqueous phase was thoroughly extracted with ether and the product mixtures were analysed by vapor phase chromatography (flame ionization detector, carbowax 10%, relative response ratio method). The results are summarized in Table 1. Reduction of enone 1 by sodium borohydride in water gives a mixture of 1,2-reduction product 2 (major) and 1,4-reduction product 3 (minor). The presence of α -cyclodextrin favours the 1,2-reduction in every cases whereas β -cyclodextrin favours the 1,4-reduction. Replacement of cyclodextrins by the same molar equivalent of glucose or α -methyl glucoside has no influence on the reactions. Cyclodextrins have no effect on the diastereomeric composition of saturated alcohols 3b and

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 $\underline{3c}$. For instance, $\underline{3c}$ is a mixture of 84% cis and 16% trans 3-methylcyclohexanol. The extension of the study is now in progress. We are now investigating the nature of enone-cyclodextrin inclusion complexes by NMR.

Table 1. Reduction of enones with sodium borohydride in water

Enone	Host	Relative y	vields/%a)	1,2/1,4 Reduction
		<u>2</u>	<u>3</u>	ratio
<u>la</u>	none	54	46	1.2
	α – CD	66	34	1.9
	β−CD	32	68	0.5
<u>1b</u>	none	78	22	3.5
	α -CD	90	10	9.0
	β-CD	65	35	1.9
<u>1c</u>	none	74	26	2.8
	α-CD	94	6	15.7
	β-CD	68	32	2.1

a) Averages for triplicate runs. The conversion of enones is superior to 95%.

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